

- Spinomar[®]NaSS -

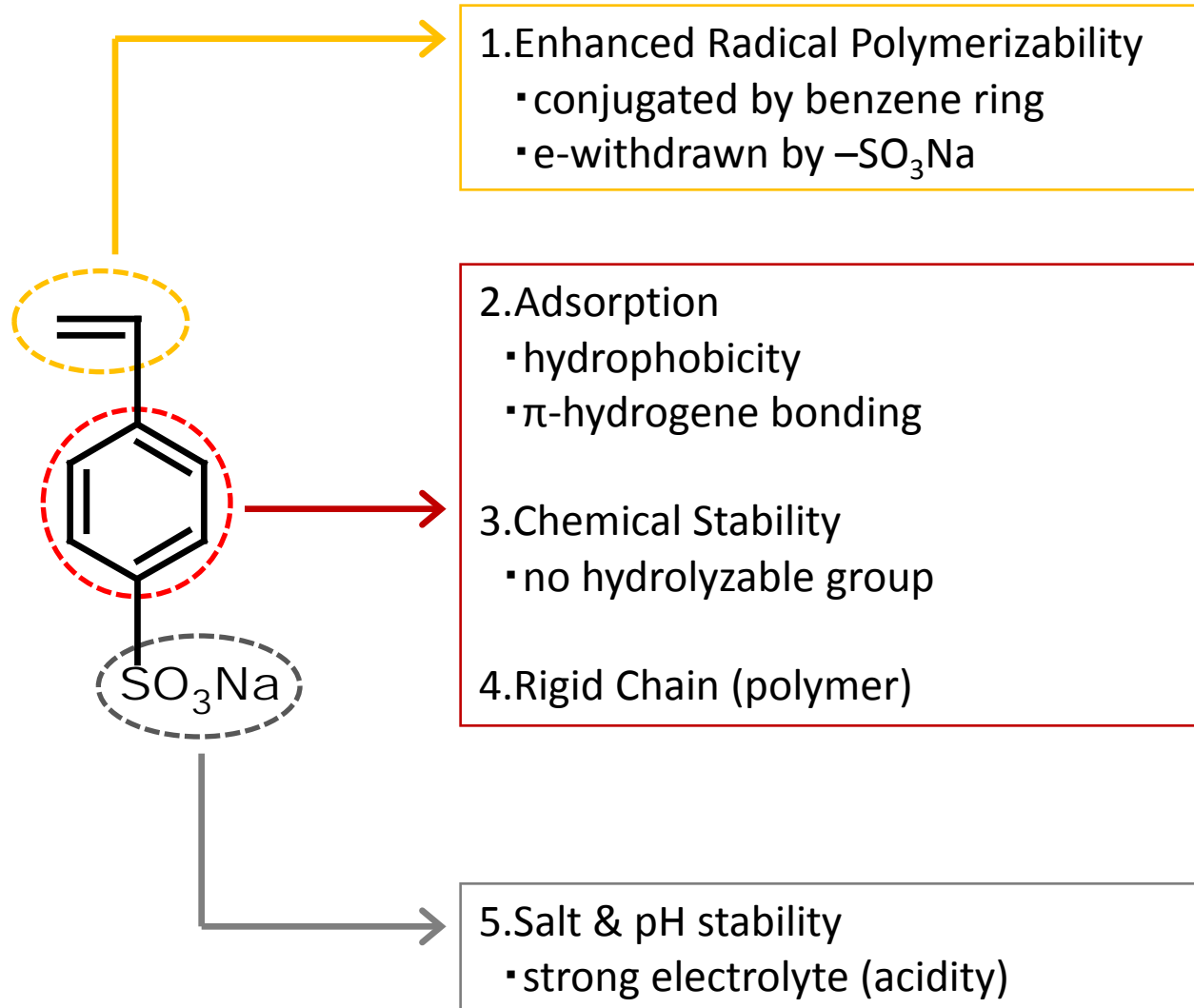


TOSOH FINECHEM CORPORATION

R&D

S.Ozoe

Properties of (poly)NaSS



Decomposition Temperature of Poly-NaSS

- Better Thermal Resistance -

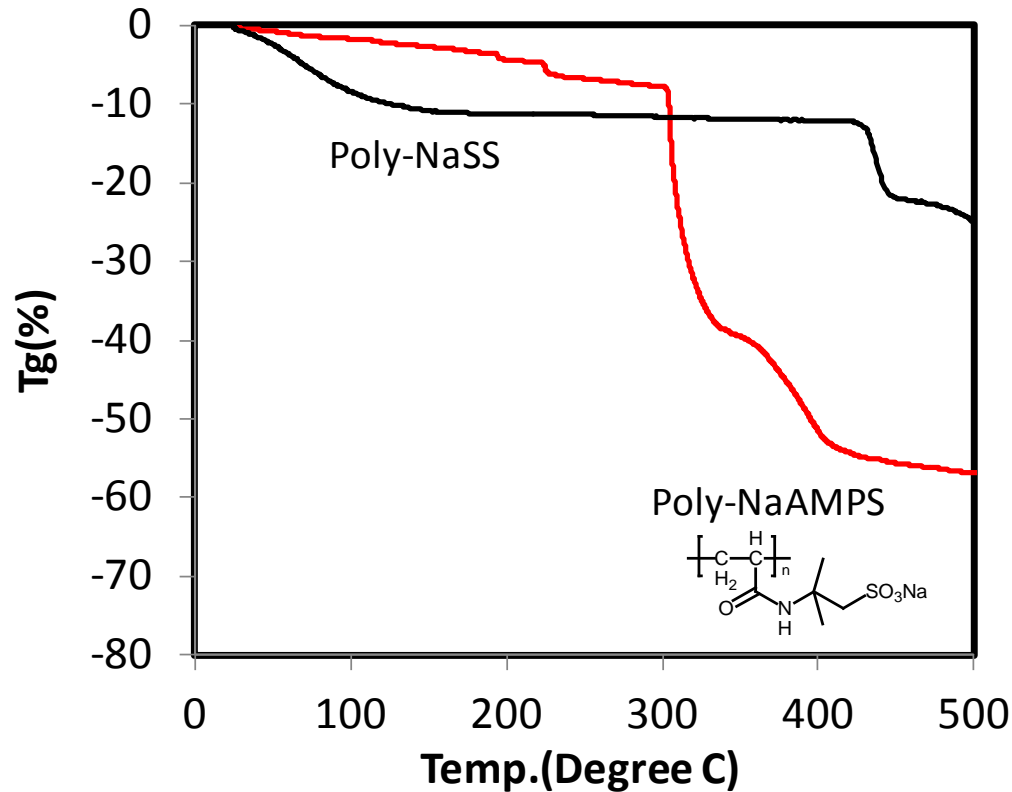


Fig. TG-DTA profile
(under N₂, 10°C/min)

H₂O₂ resistance of Poly-NaSS

- Better Resistance -

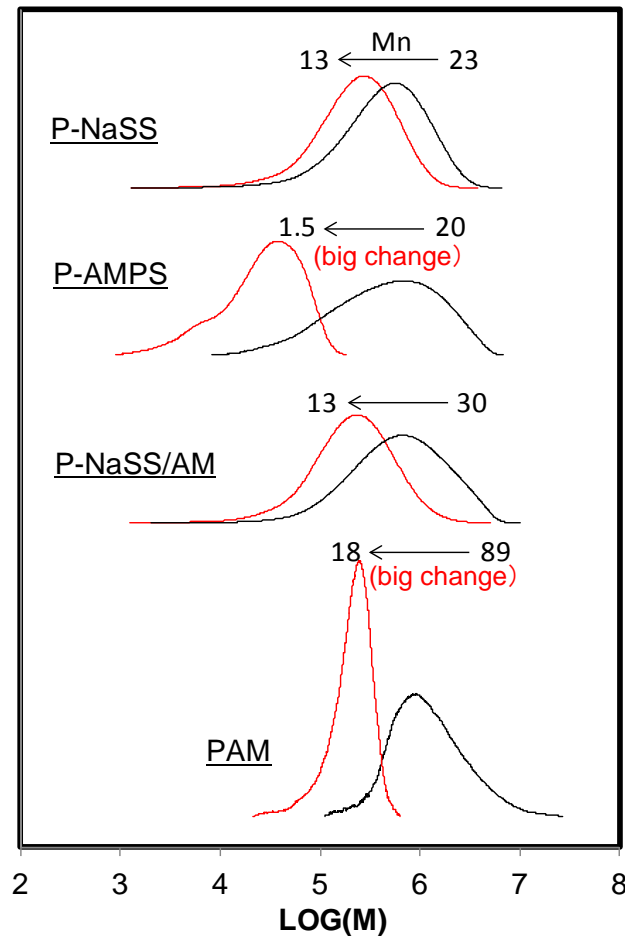
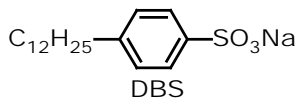
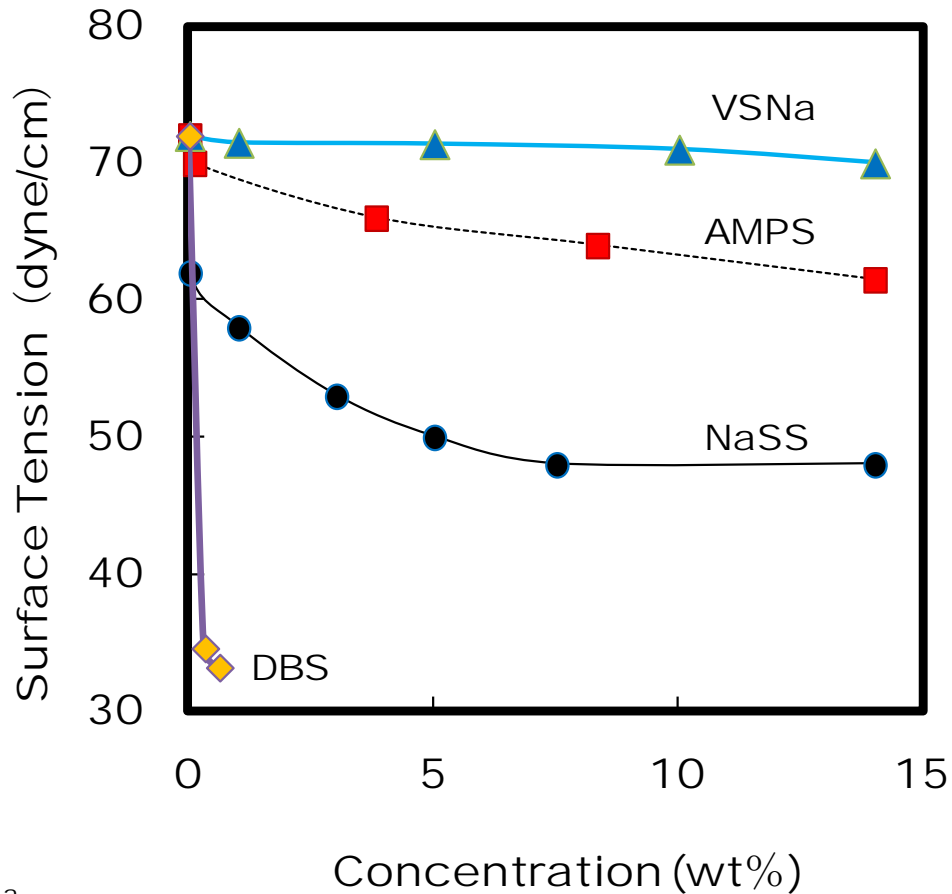


Fig. Mw(GPC) Distribution Change

(20% polym. aq.=100g/30%H₂O₂aq.=5g → 50°C × 14days)

Surface activity

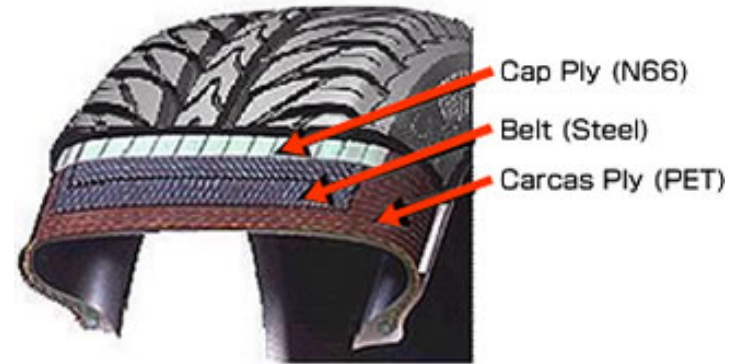
- More surface active -



Main Application : Emulsion (Polymerization)



Emulsion Paint



Tire Cord/Rubber Adhesive



Acrylic Fiber



Remedy for teeth

Needs for emulsion paint

◆ Needs (especially for architecture)

- Quick dry by high solid (filler) \Leftrightarrow High colloidal stability
- Water resistance
- Adhesion



- Surfactant system is the Key

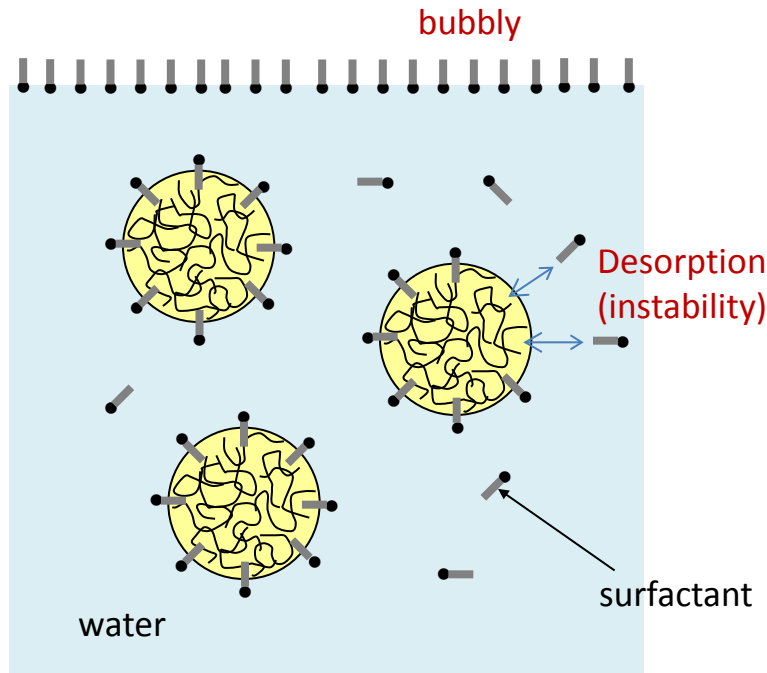
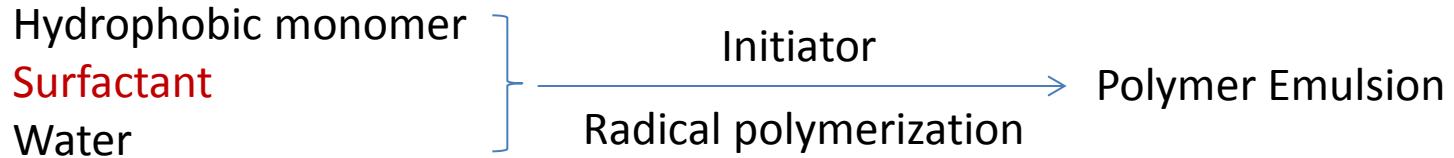
◆ Reactive surfactant

- 10-20% of conventional surfactant has replaced by reactive one in Japan and 1-2% in other countries.
(The Chemical Daily of Japan, Nov.2016)

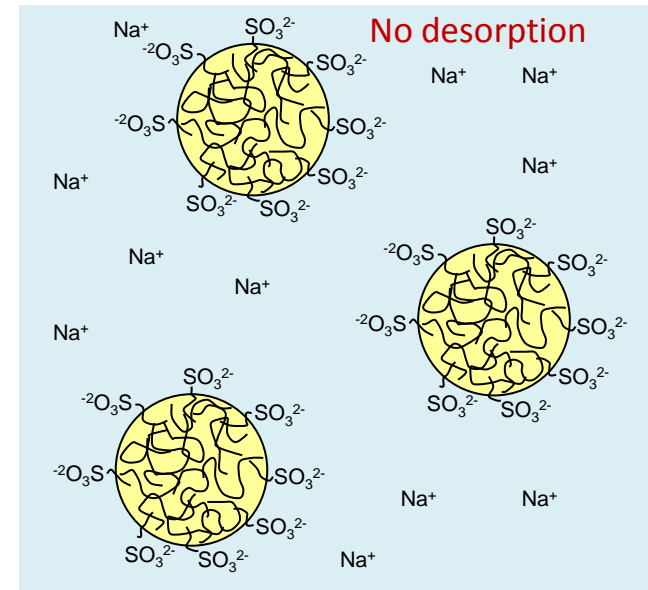


◆ Room for growth

Emulsion stabilizing system

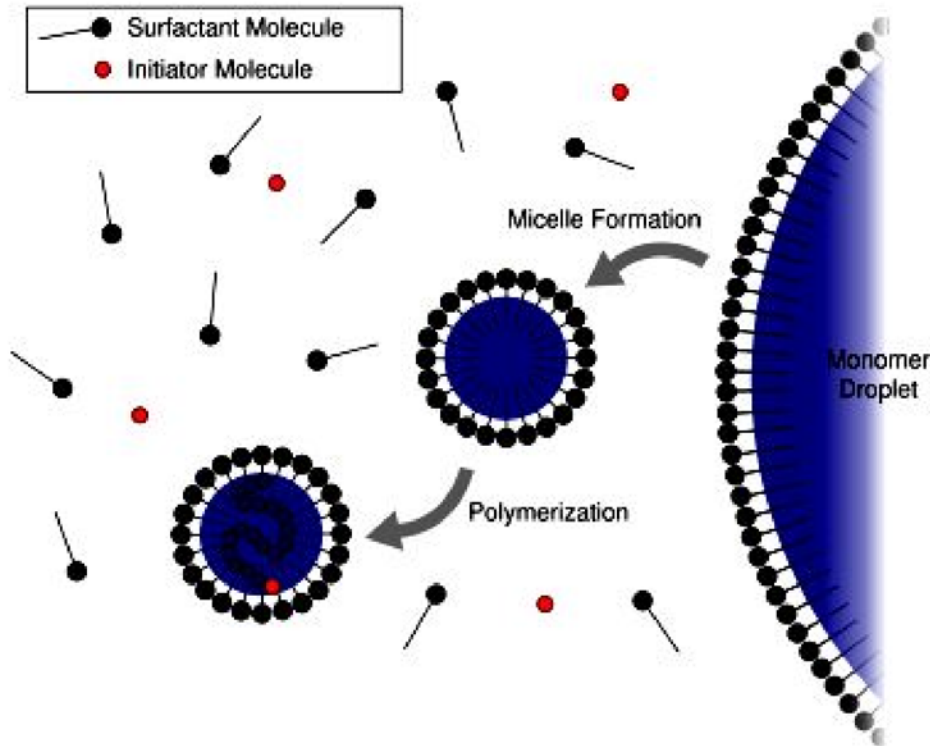


Stabilized with conventional surfactant



Stabilized with reactive surfactant

Emulsion Polymerization

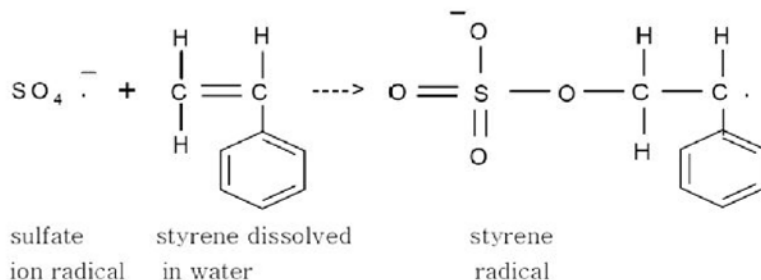


Features

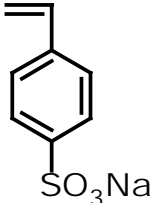
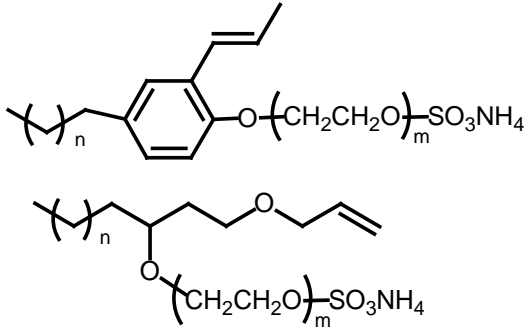
- Heterogeneous system
- Nucleation site: water (in micellar or homogeneous)
- ~20nm – ~1μm
- Monodisperse
- Polymerization time: fast
- High MW
- Surfactant-free is possible

Components

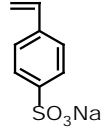
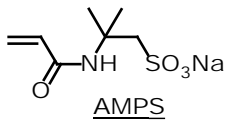
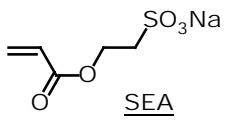
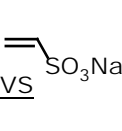
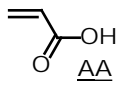
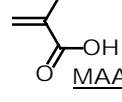
- Monomers
- Water
- Water-soluble initiator
- Surfactant



Advantages of NaSS (1/2)

	NaSS	Example of Reactive Surfactant
		
Surface Activity	Good	Excellent
Polymerizability	Excellent	Poor
Heat Stability	Excellent	Poor
Other	Compact Molecule	<ul style="list-style-type: none"> · decrease adhesion · buried into particle during Em.polym.

Advantages of NaSS (2/2)

	Sulfonate Type				Carboxylate Type	
						
Acidity (pH stability) (Salt stability)	High (Strong Electrolyte)				Low (Weak Electrolyte)	
Hydrolytic Stability	Excellent	Fair	Poor	Excellent		
Heat Stability	Excellent	Good				
Polymerizability	High			Low	High	
Surface Activity	Good	Poor				
Solubility	Poor	Fair			Good	



Favorable for Emulsion polymerization **but.....**

Points to Note

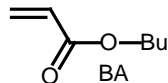
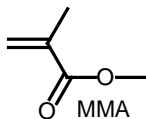
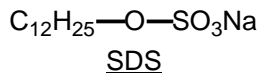
1. NaSS sometimes reacts too fast in water phase and forms water soluble polymer (= decrease water resistance).
→ dosing condition, initiator system
2. NaSS incorporation changes significantly by initiator system, monomer/surfactant composition and dosing condition.
3. Excess Na sulfonate decreases water-resistance of polymer film.
→ appropriate quantity
4. NaSS incorporation is limited (<5%) due to its oil insolubility.

Colloidal stability increases by introduction of NaSS

[Quoted from Jose M.Asua : European Polymer Journal 93 (2017) 480-494]

MMA/n-BA=1/1wt.r

NaSS (wt%/MM)	Basic Characteristics of Latex				Salt Tolerance [Coagulation(%)]			Freeze Thaw Stability [Z-ave change(%)]	
	-SO ₃ Na (mmol/kg)	Surface Tension (mN/m)	ζ-potential (mV)	Z _{-ave} (nm)	0.02M CaCl ₂	0.05M CaCl ₂	1M NaCl	1st cycle	3rd cycle
3.6	86	51.8	58.4	281	<6	<6	<6	<6	
1.3	31	54.8	52.8	241	<6	massive coagulation	massive coagulation		
1.0	24	58.1	52.6	251	<6				
0.5	12	63.3	51.0	250	<6				
SDS=1.2	21	40.9	46.6	255	19				



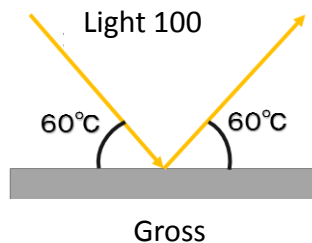
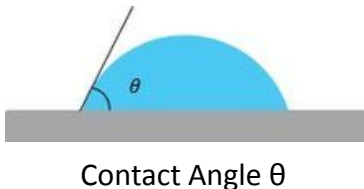
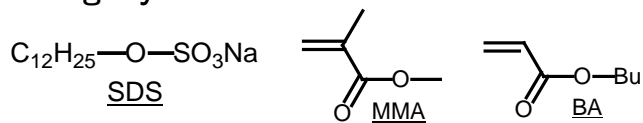
NaSS stabilized emulsion gives clean film surface

[Quoted from Jose M.Asua : European Polymer Journal 93 (2017) 480-494]

MMA/n-BA=1/1wt.r

NaSS (wt%/MM)	Contact Angles			Moisture Permeability (g·mm/m ²)
	Before Washing	After Washing	Gloss at 20°	
3.6	32.5	—*	66.4	-
1.3	68.4	69.4	64.3	59
1.0	71.1	69.7	69.1	14
0.5	69.8	69.9	66.1	10
SDS=1.2	52.4	68.3	56.3	11

* Integrity lost



Surface is covered with SDS
(=white area)

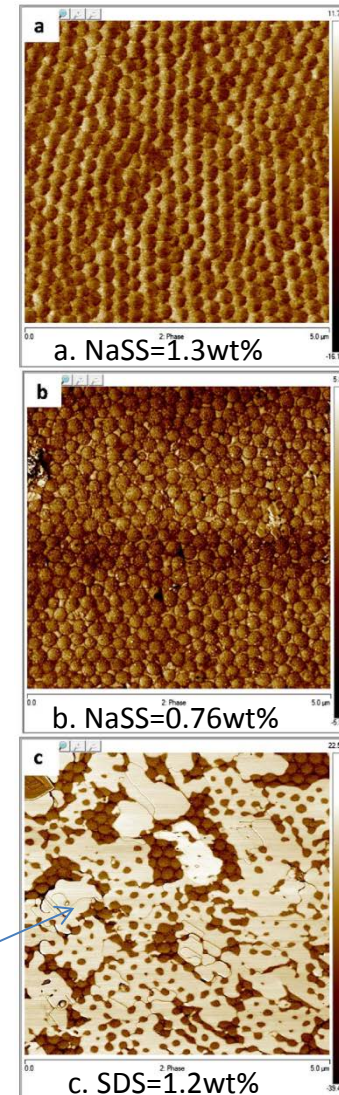


Fig. AFM phase images of film-air interface

Clean surface gives better adhesion

[Quoted from M.Okubo : J. Adhe.Soc.Jap. Vol.18(4),1982,153-158]

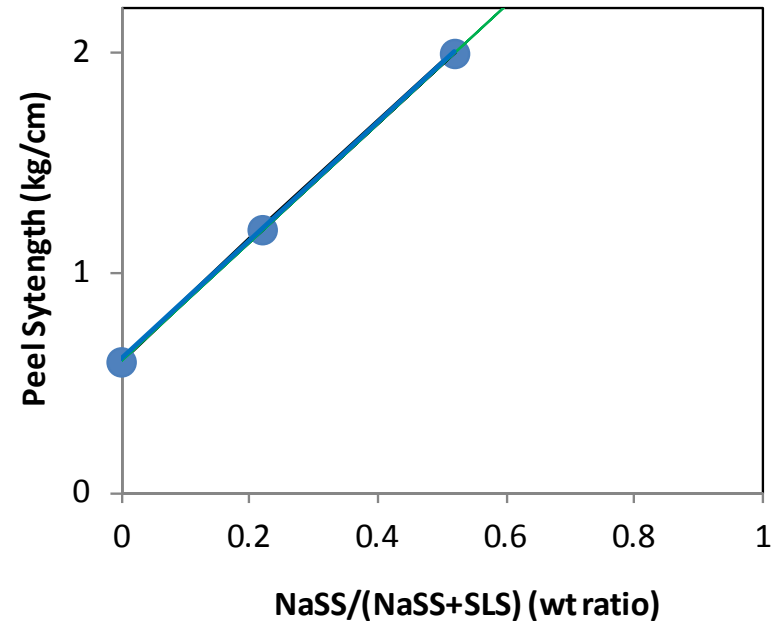
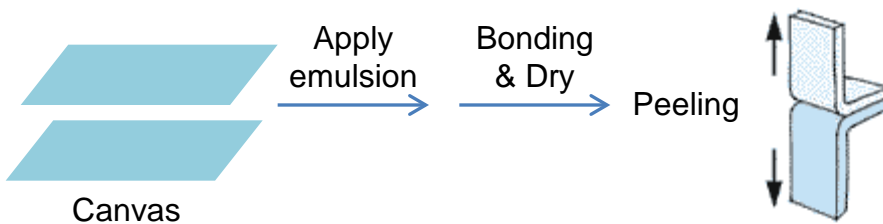
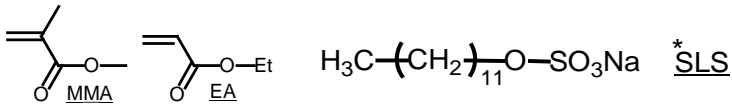
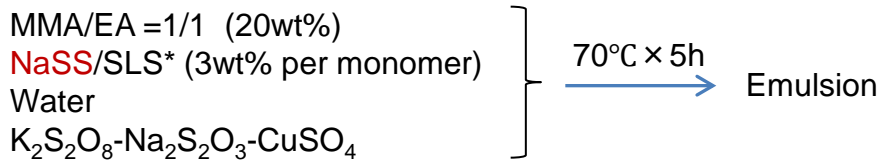


Fig. Effect of NaSS on Peel Strength

Water resistance improves by NaSS dosing condition

[Quoted from M.Okubo : J. Adhe.Soc.Jap. Vol.18(12),1982,530-535]

Initial Dose

St/n-BA=1/1 wt.r
(A) NaSS=1mol%×100 ~ 0%
Deionized Water
Azo-Initiator

2nd (Stepwise) Dose

(B) NaSS=1mol%×0 ~ 100%

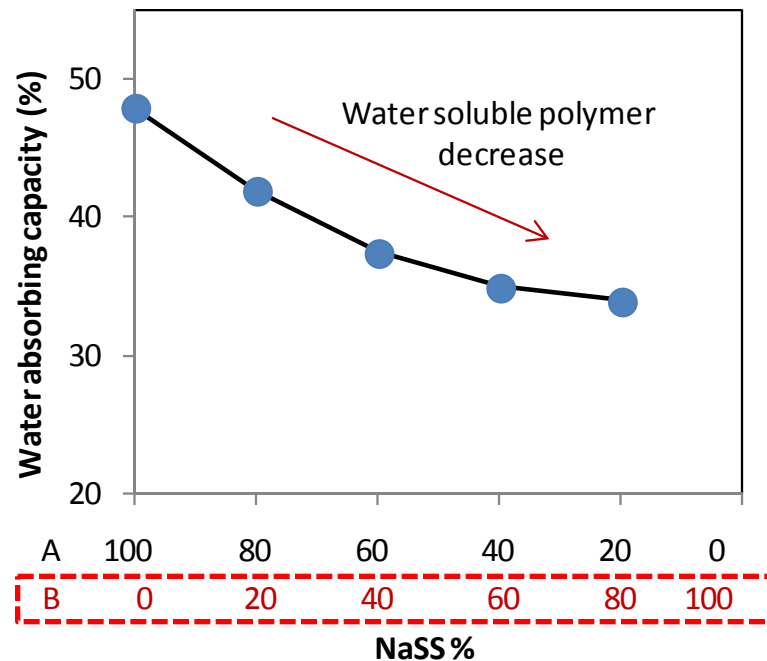
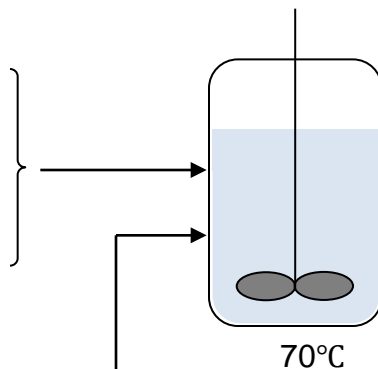


Fig. Effect of NaSS dosing condition on the water absorbing capacity of film(30°C × 72h)

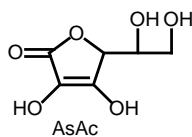
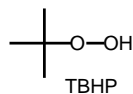
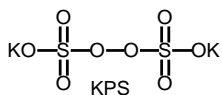
NaSS incorporation is affected by initiator system

[Quoted from Jose M.Asua : Polymer 117 (2017) 64-75]

MMA/n-BA=1/1wt.r

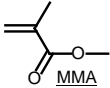
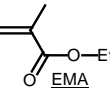
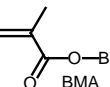
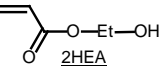
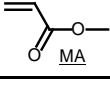
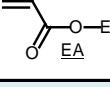
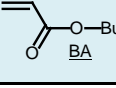
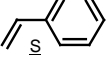
Initiator System	Basic Characteristics of Latex			Salt Stability [Coagulation(%)]		Freeze Thaw Stability [Z-ave change(%)]	
	NaSS Incorporation (%)	Particle Size(nm)	Coagulation (wt%)	0.02M CaCl ₂	0.5M NaCl	1st cycle	3rd cycle
KPS	60.7	265	1.9	Good	NG	<6	<6
H ₂ O ₂ /AsAc	64.1	250	0.7	NG		NG	NG
TBHP/AsAc	72.5	240	0.7	<6	<6	<6	<6

NaSS=1.3wt%/total monomer



Hydrophilic comonomer increases NaSS incorporation

[Quoted from Jose M.Asua : RSC Advances 2016,6,63754-63760]

Monomer	Solubility in Water (mM) at 25°C	Particle Size (nm)	NaSS incorporation(%)
 MMA	150	188	75.5
 EMA	45	196	57.2
 BMA	4	201	53.8
 2HEA	∞	476	99.7
 MA	650	193	86.9
 EA	150	194	77.6
 BA	11	211	48.7
BA/2HEA=95/5mol%		193	78.9
 Styrene	3.5	197	42.3

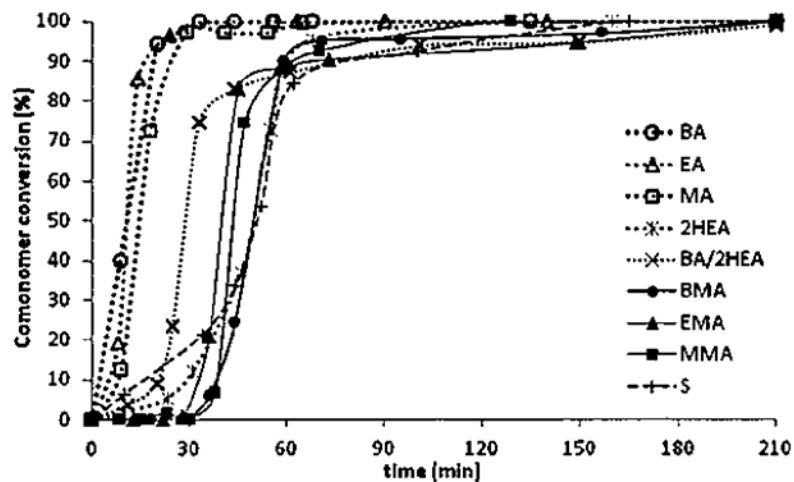
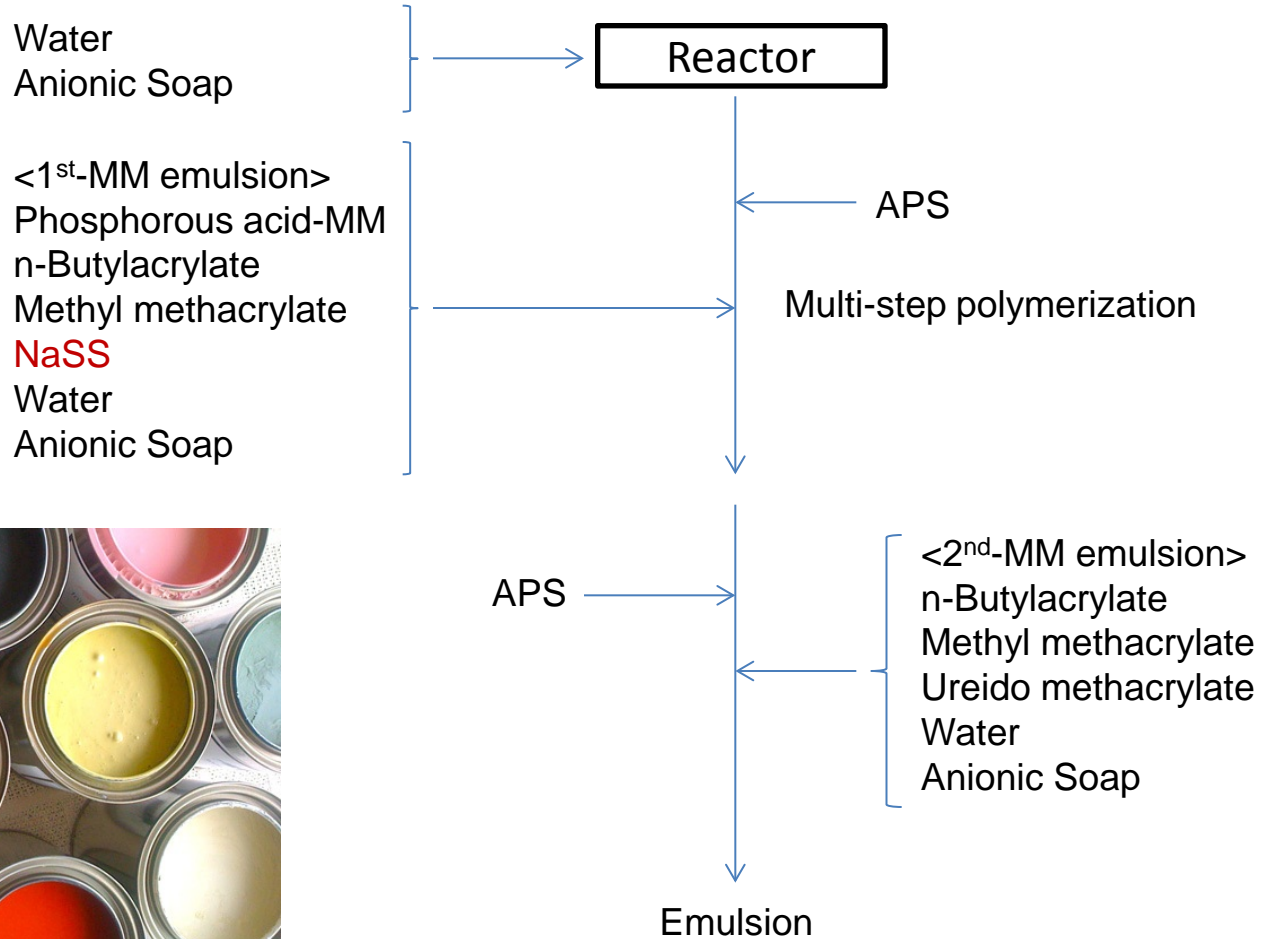


Fig. Time vs Monomer Conversion

Emulsion Paint

(US2012/58277)

★Excellent Colloidal Stability



Acrylic Fiber : Dyeing Site

(JP57-10613)

- ★ Distribution of $-SO_3Na$ group
- ★ Spinning behavior (less void)



Monomer Emulsion

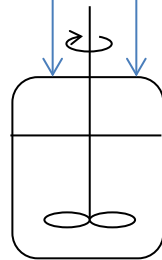
Acrylonitrile
Vinylchloride

NaSS*

(*stepwise dosing)

Radical Initiator

Other Additives



Salting out

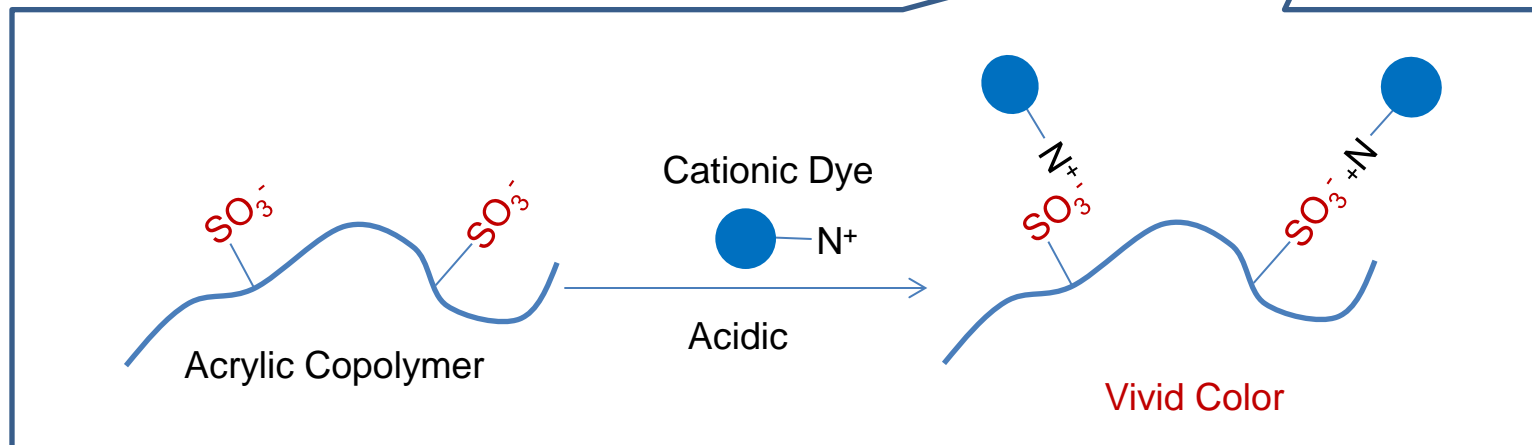
Dissolving

Spinning

Acrylic Fiber

Emulsion Polymerization

Dyeing



Acrylic Fiber

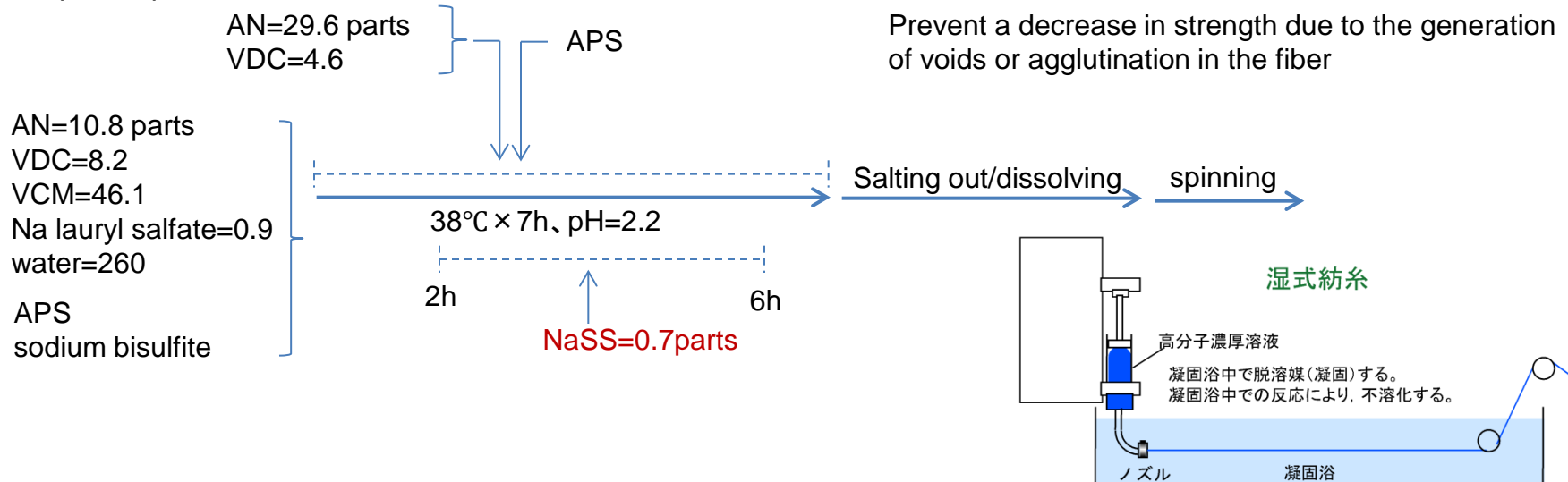
- Strong demand for flame retardency of acrylic fiber
 - addition of flame retardant → decreasing properties, drop out of retardant
 - copolymerization of VCM/VDC → decreasing transparency
 - copolymerization of NaSS → easy for AN \geq 85%, hard for modacrylic



- Improvement of **NaSS feed condition (=decreasing void)** (JP58-91710)

AN=40~65wt%
 VCM or VDC=31~59.9wt%
 NaSS=0.1~4wt%
 surfactant=0.2~5wt%

(example recipe)



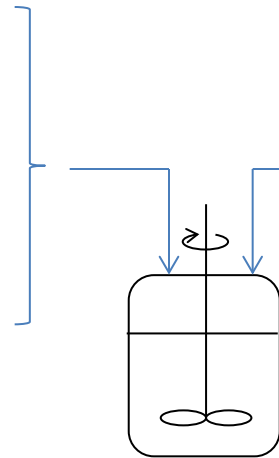
Latex Adhesive for tire cord(PET)

(JP4786061)

★Excellent
Colloidal & Heat Stability
Adhesion (130 ,170°C curing)

Monomer

NaSS (0.1)
2-Vinylpyridine (7.5)
Styrene (35.1)
Butadiene (16.5)
Glycidyl methacrylate (0.9)
t-dodecylmercaptane (0.2)



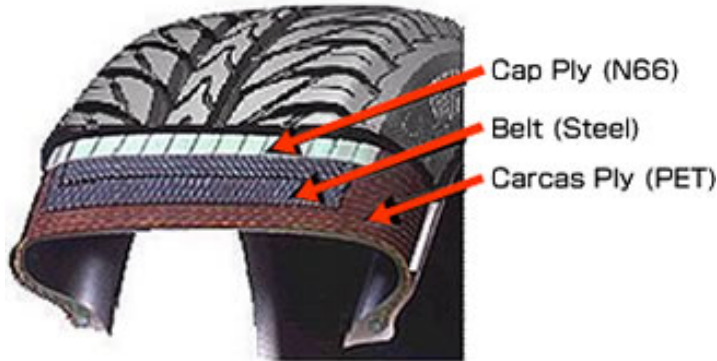
Water (140)
K-rosinate (1.5)
Dispersant (1.0)
APS* (0.3)

Conv.=80-90%

Formulation

Adhesive

K-rosinate (2.0)
2-Vinylpyridine (6.0)
Styrene (6.0)
Butadiene (28.0)
t-dodecylmercaptane (0.4)



Tire=Critical Safety Parts

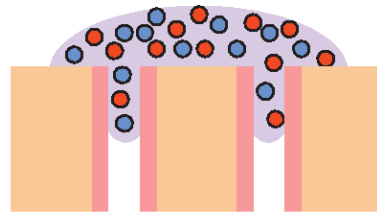
(*Ammonium persulfate)

Hyperesthesia-suppressing Agent

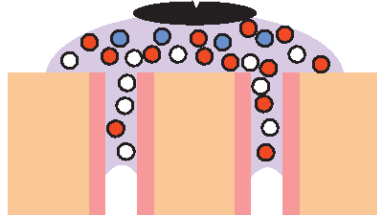
Protective coating using NaSS copolymer particle



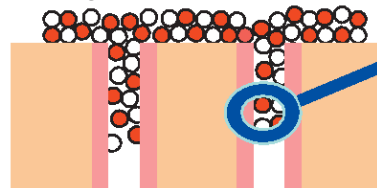
Coating



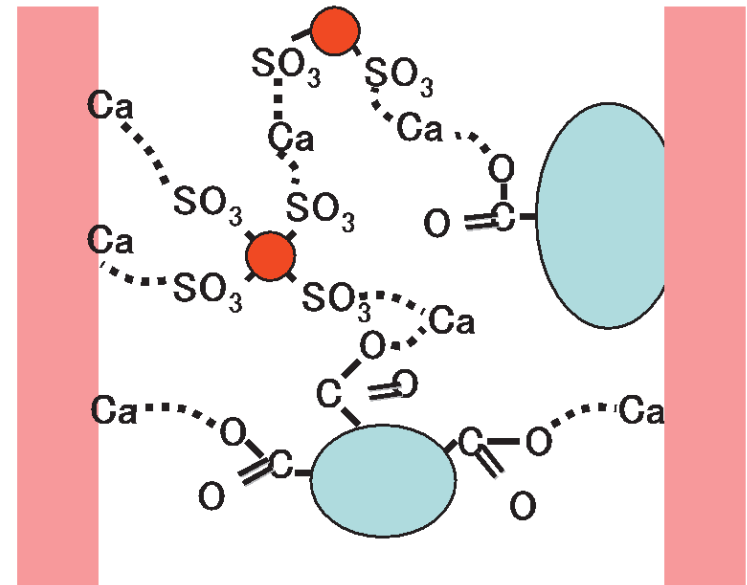
Rubbing



Drying



- ★ Colloidal Stability
- ★ Adhesion
- ★ Low Toxicity



- oxalic acid
- calcium oxalate
- MMA/NaSS copolymer particle
- peritubular dentin
- intertubular dentin

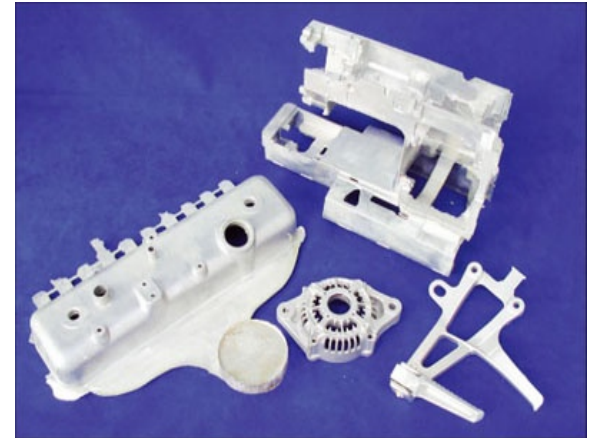
Dispersion/Dispersant



Chlorinated PVC



Agrochemical Granule



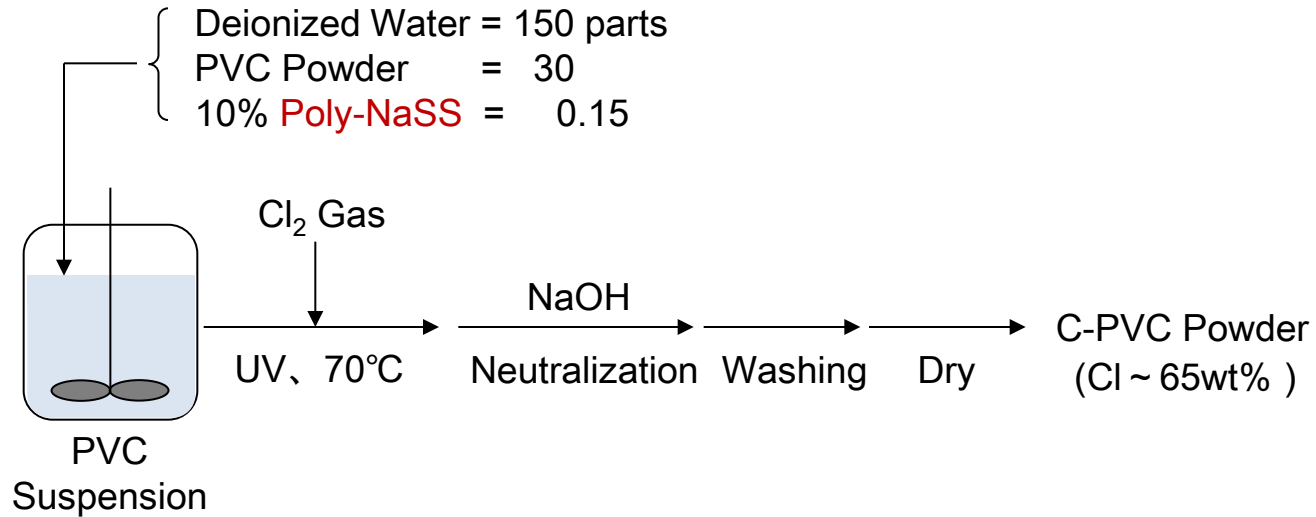
Mold Lubricant



Scale Inhibitor

Chlorinated PVC

(JP3176504)



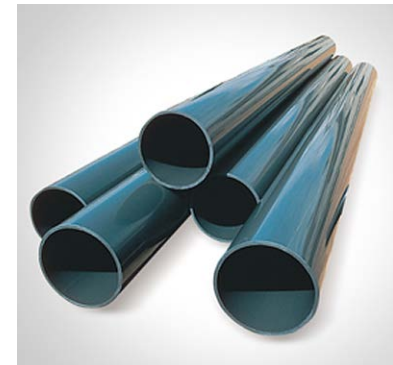
Heat distortion temperature(°C)

PVC = 60 ~ 80

C-PVC = 80 ~ 120

★ Stable Dispersant

Poly-NaSS improves uniformity of the chlorination reaction.
As the result, thermal stability (=color fastness and transparency)
of C-PVC improves.



Water dispersible agrochemical granule

(JP63-236701)

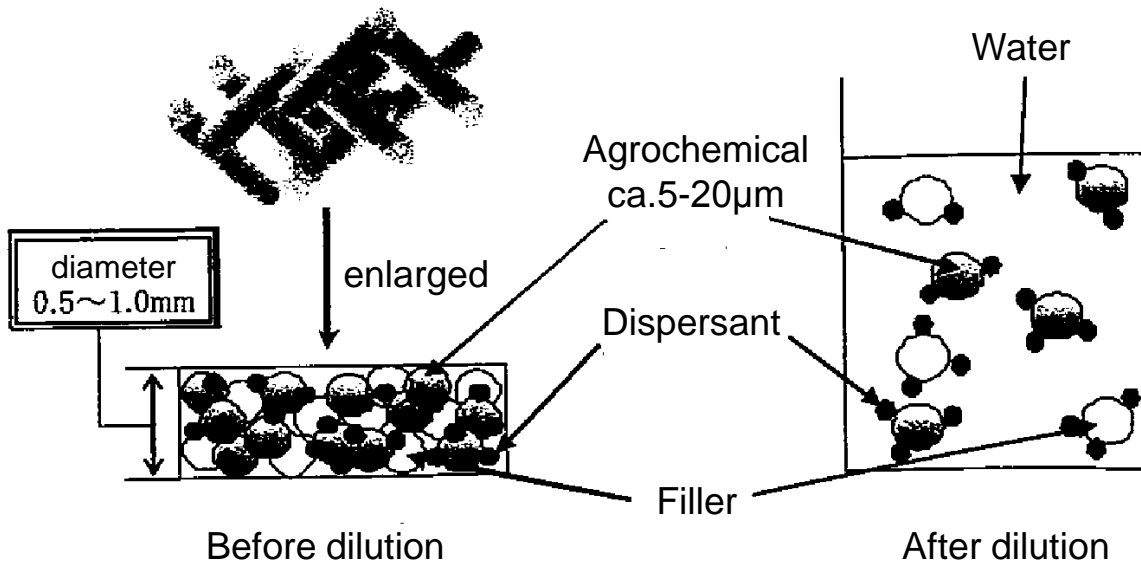
★ Dispersant & Binder

Agrochemical
Kaolinite clay
Poly-NaSS
Other additives
Water

granulation

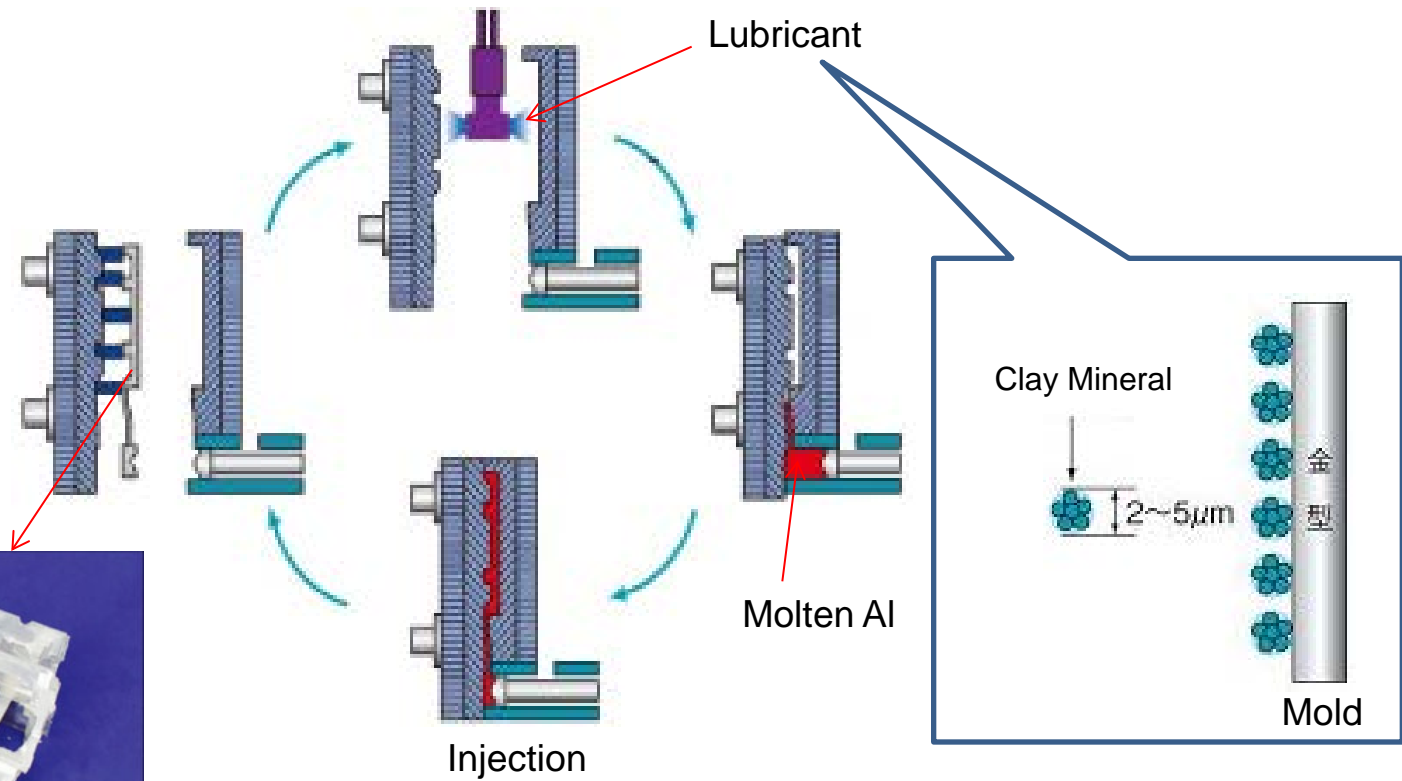
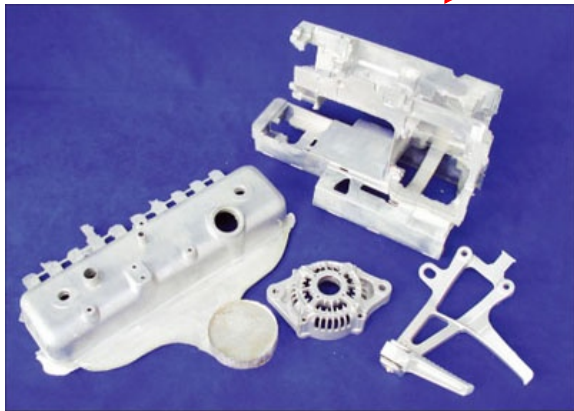
fluidized bed drying

product



Water-based Mold Lubricant

(JP4464214)



★ Heat stable dispersant & Adhesion

Poly-NaSS enhances dispersion stability and adhesion (wettability) to the mold.

Polymeric Surfactant (dispersant)

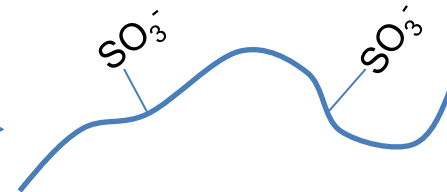
★ Dispersant & π -hydrogene bonding

NaSS

Co-monomer

Hydrophilic MM
Hydrophobic MM

Radical Polymerization



Durable to Ca^{2+} , Mg^{2+} etc
Good heat resistance

<Example Use>

Scale inhibitor (Water Treatment, Oil Field)

Sizing emulsion for paper (durable to hard water)

Dispersion of pigment, silicate, CNT etc

Washing agent, Thickener



Calcium phosphate inhibitors (1/2)

[Quoted from Zahid Amjad : Phosphorous Research Bulletin,20,165-170(2006)]

★ NaSS moiety increases the heat stability of polymer.

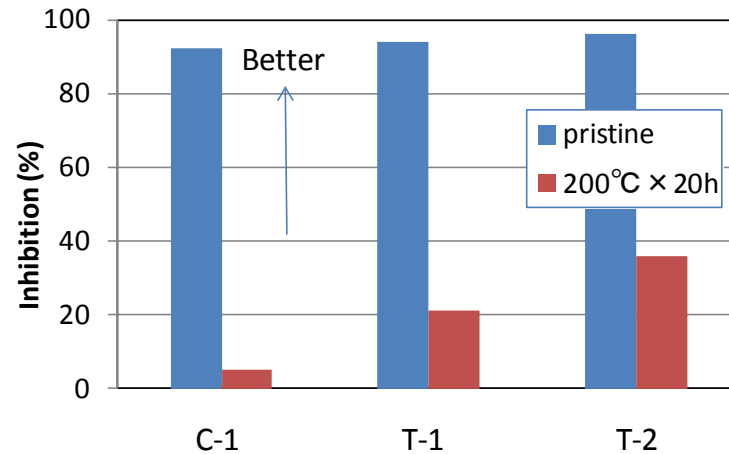
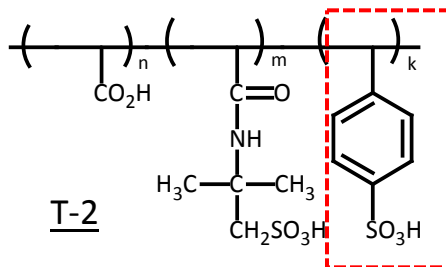
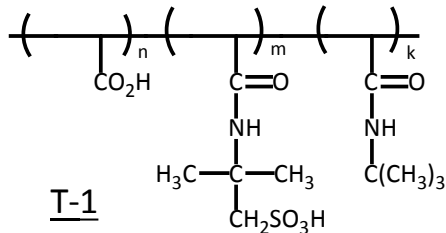
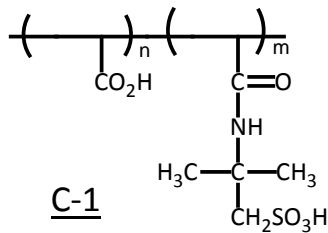


Fig. Effect of heat treatment on the polymer performance

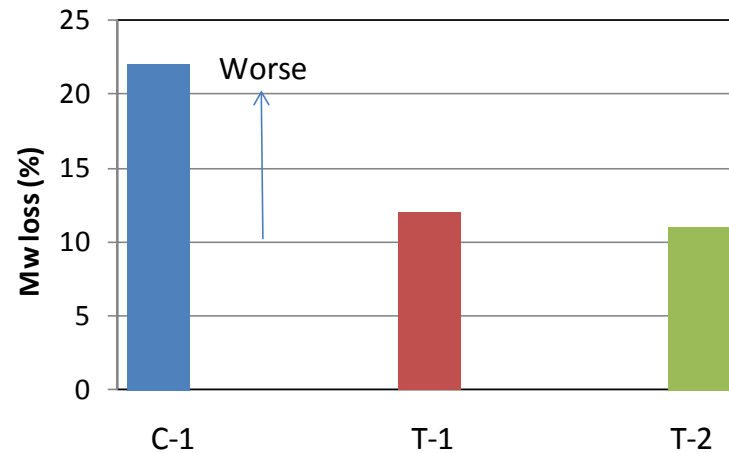


Fig. Mw loss of polymer after heat treatment

Calcium phosphate inhibitors (2/2)

[Quoted from Zahid Amjad : Phosphorous Research Bulletin,20,165-170(2006)]

★ NaSS terpolymer maintains inhibition performance after heat treatment.

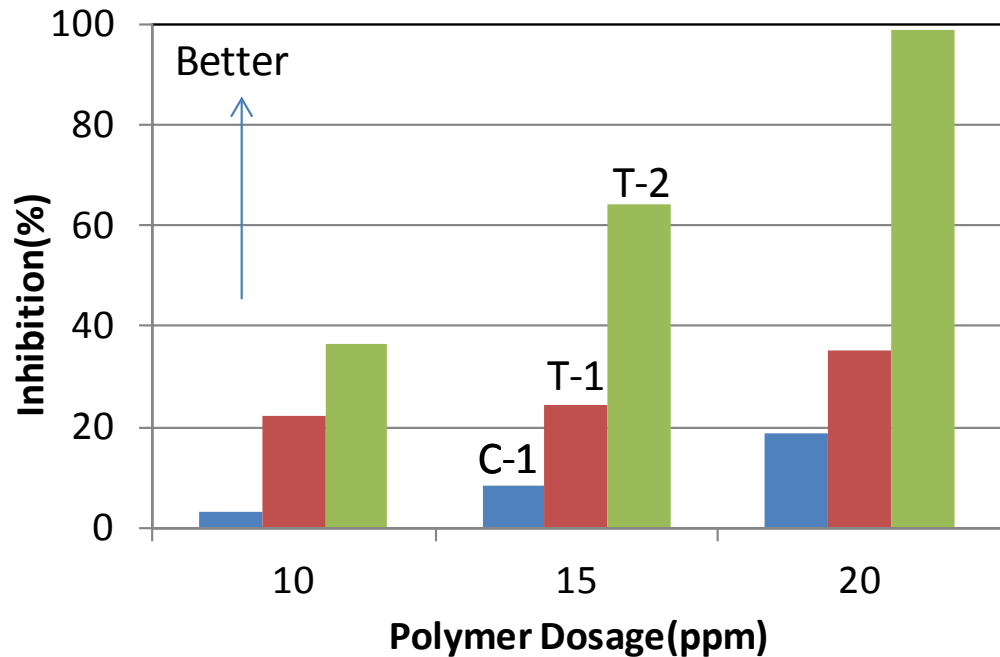
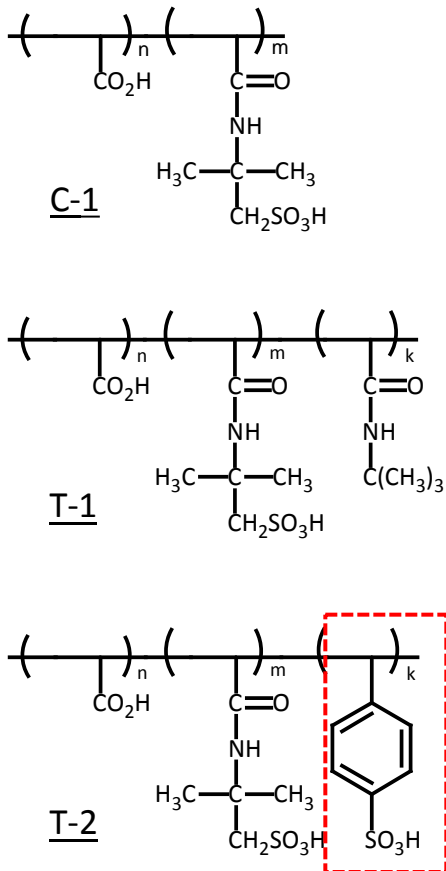
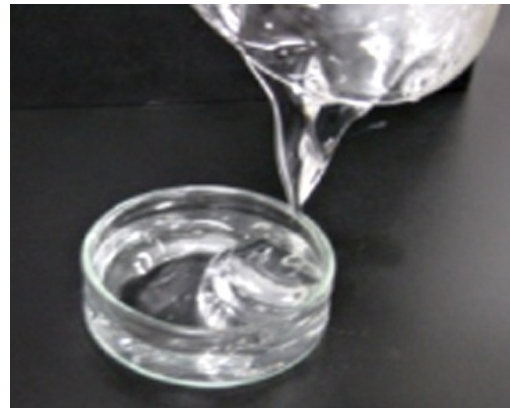


Fig. Effect of polymer dosage on the heat-treated sample

Toiletary



Ironing Aid



Thickener



Textile Washing

Advantage of POLYNASS[®]

★Low APHA(=good hue): protected by our Patent**

MA-2005L*
(APHA80)

PS-1
(>250)

PS-5
(>250)

PS-50
(70)

PS-35*
(30)

PS-100
(70)



*for specific customer

** (JP5946094), US9505713, TW1573781

Ironing Aid

★ Poly-NaSS acts as heat-stable glue, dispersant



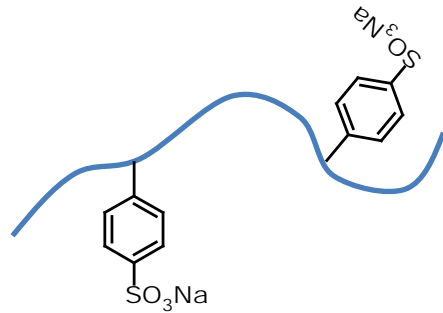
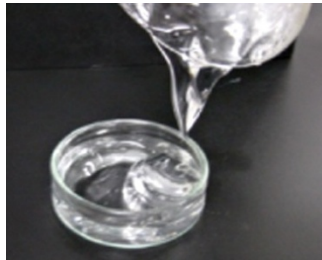
Poly-NaSS (substitute of starch)
Silicone polymer (smoothing agent)
PPG (stabilizaer)
Preservative
Flavor
Etc
Water

Poly-NaSS has an excellent resistance to heat and discoloration.

Aqueous thickener for bleaching detergent

(JP4577308)

★ NaSS moiety increases H₂O₂ stability of Poly-AMPS



	Polymer Composition(mol%)		
	NaSS	AMPS	AA
Polymer-A	10	40	50
Polymer-B	10	70	20
Polymer-C	-	50	50
Polymer-D	-	80	20

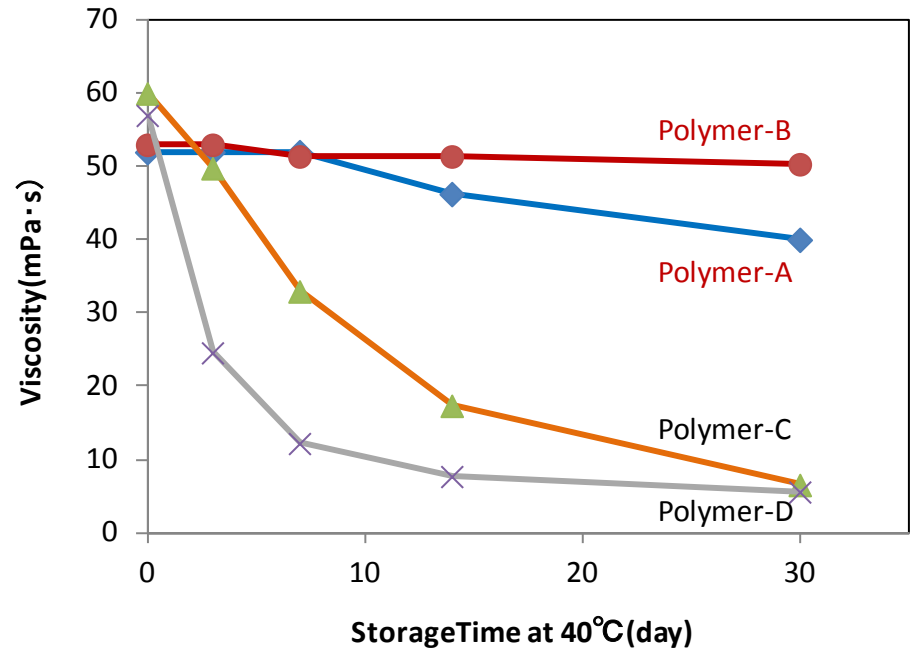
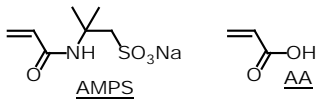
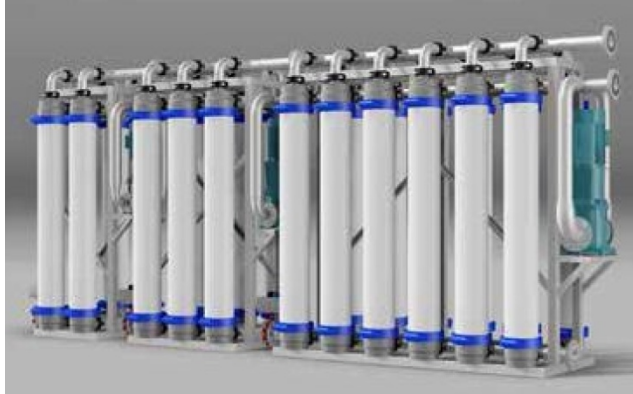


Fig. Viscosity stability of detergent composition (Citric acid=10%, H₂O₂=0.2%, pH ~ 2)

Anion/Cation Interaction (Layer-by-Layer)



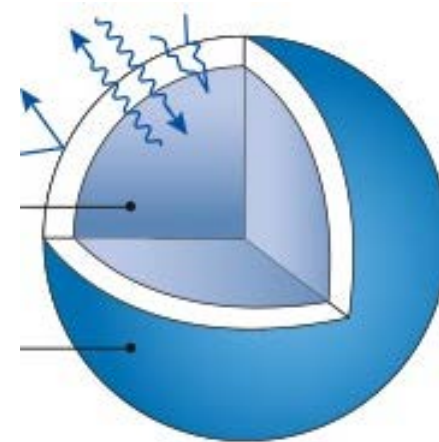
Nano-filtration



Anti-clumping





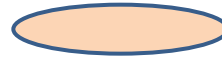








Allergen Catcher



Microcapsule

Filtration Membrane

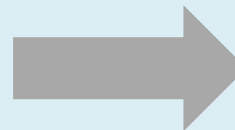
Size	1nm	10nm	0.1μm	1μm	10μm	
Object	trihalomethane  Agrochemical • Organic  ion  Multivalent ion 	Virus 	Escherichia coli Bacteria 	Bacteria 	CRYPTOSPORIDIUM 	
Type						
	RO (逆浸透)	<u>NF (ナノ濾過)</u>	UF (限外濾過)	MF (精密濾過)		

< Conventional Production method >

- Interfacial condensation
- Graft reaction using enzyme
- Radiation graft polymerization
- Chemical modification of polymer

Simplified

Decrease deficit

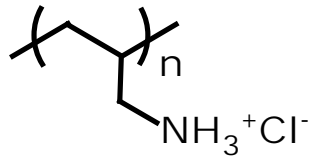


< Layer-by-Layer >

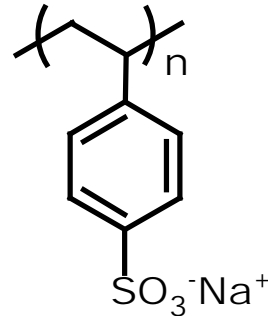
- Coating by polymer aq.

Nano-filtration by LBL technique

Very simple

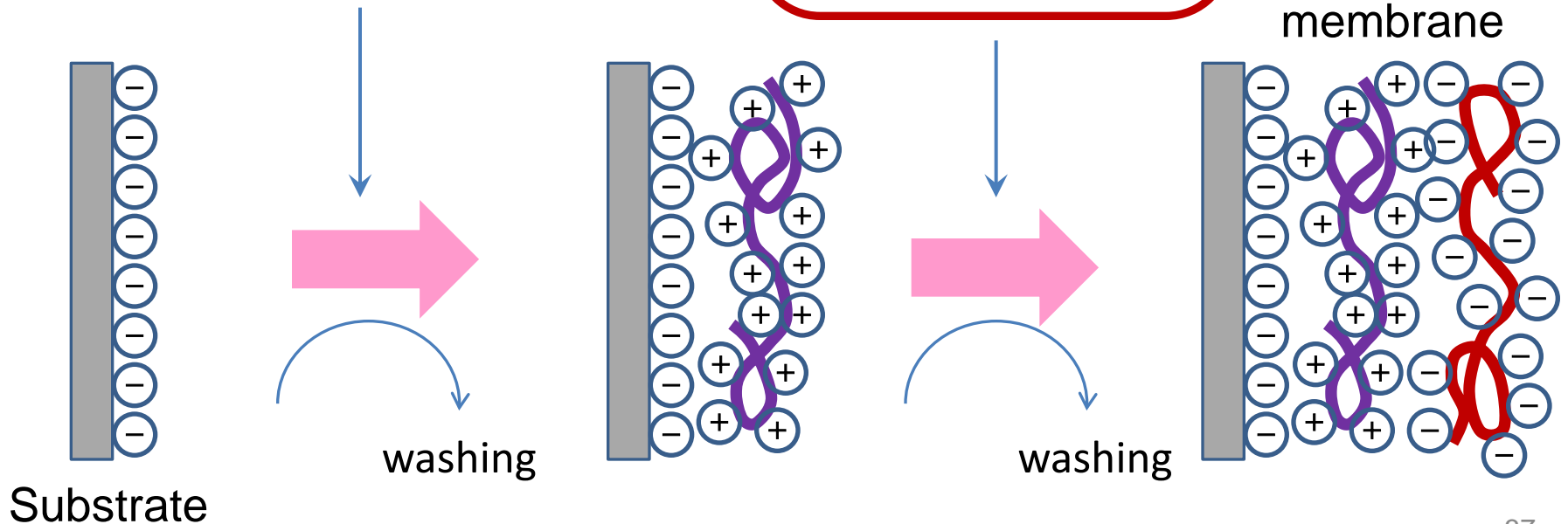


Cationic polymer aq.

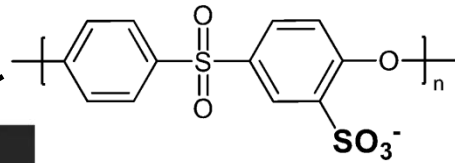


Anionic polymer aq.

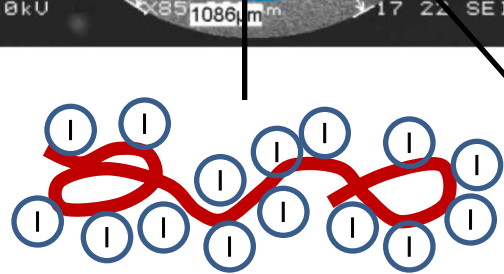
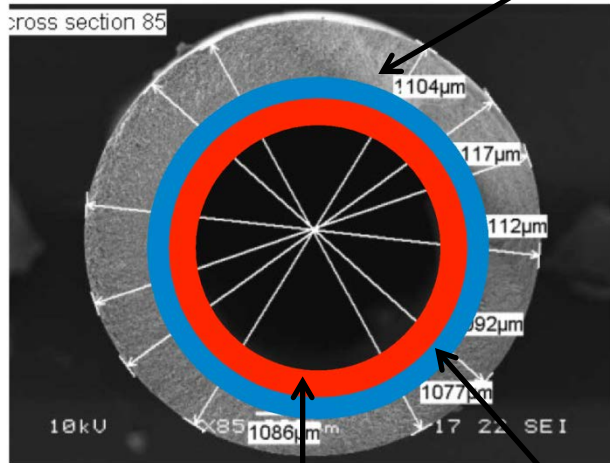
Nano porous membrane



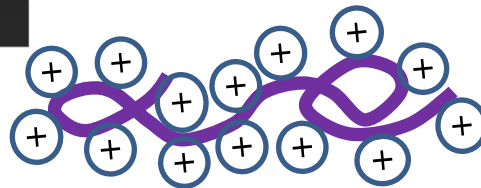
Nano Filtration Membrane



Hollow Fiber Membrane



Poly-NaSS layer

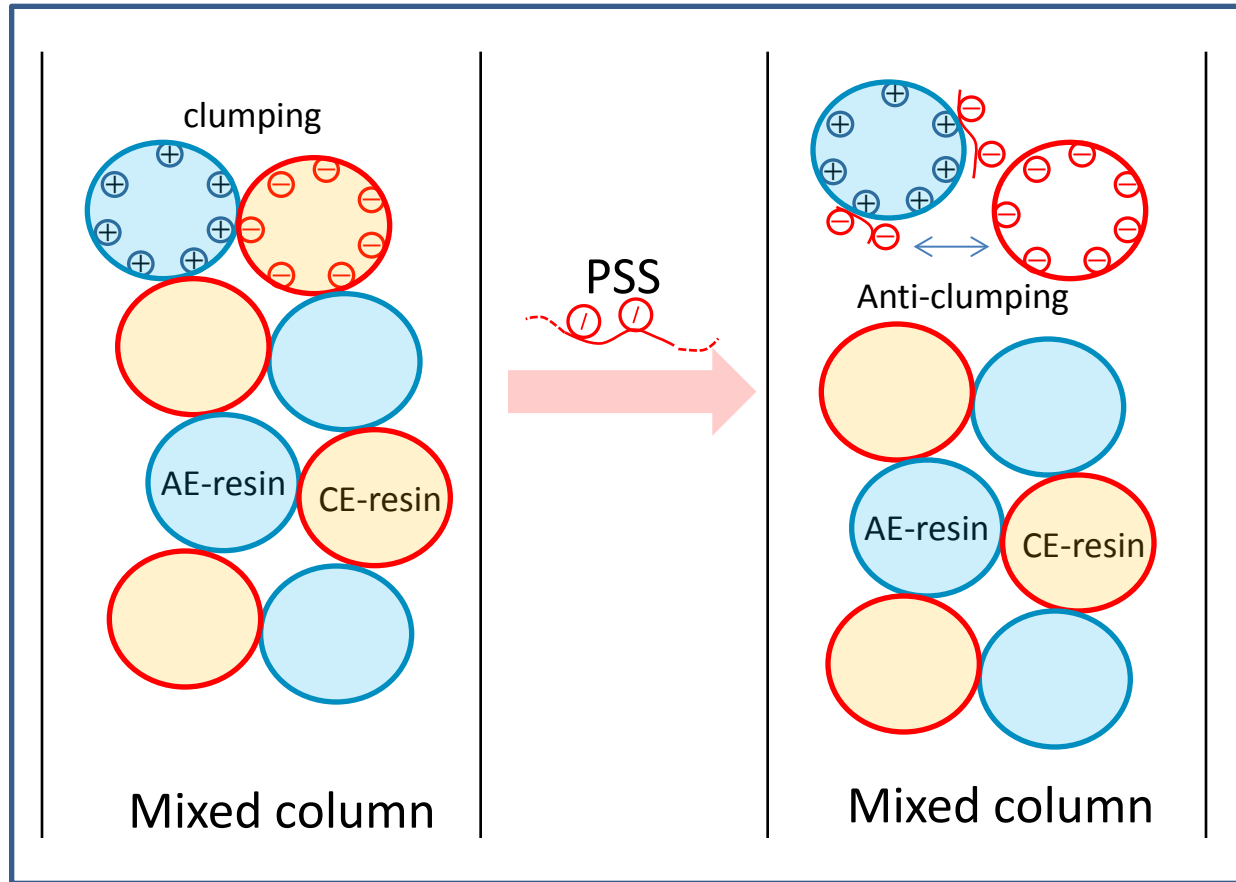


Cationic polymer layer



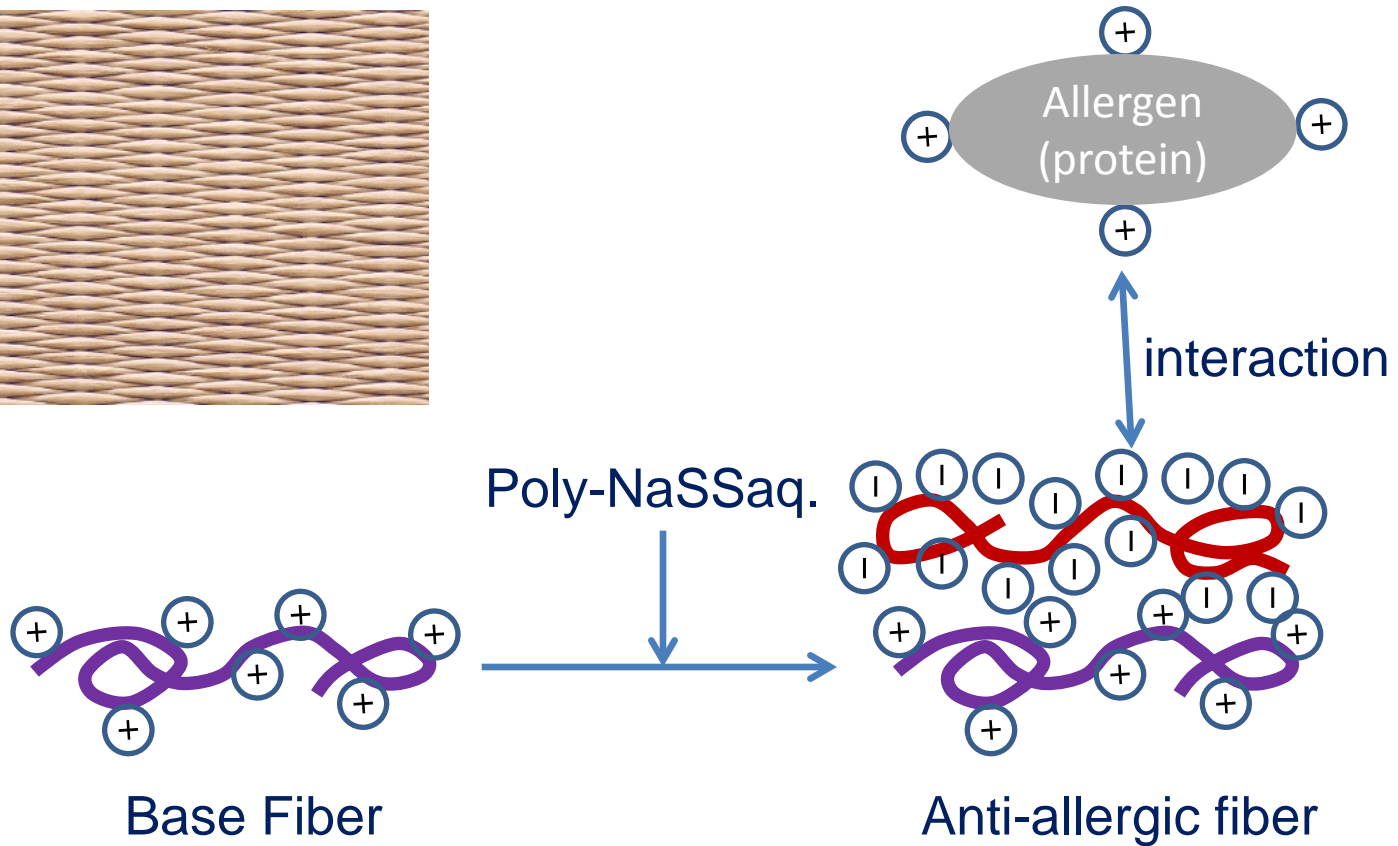
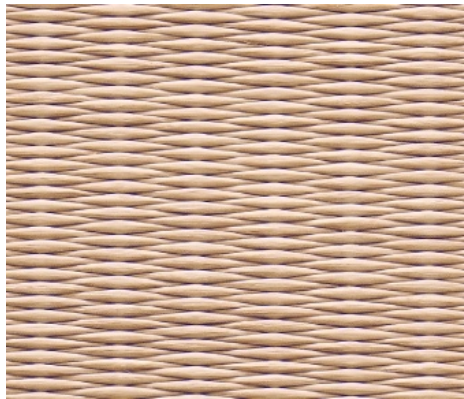
Anti-clumping of A/C-exchange resin by Poly-NaSS

By Cation-Anion Interaction



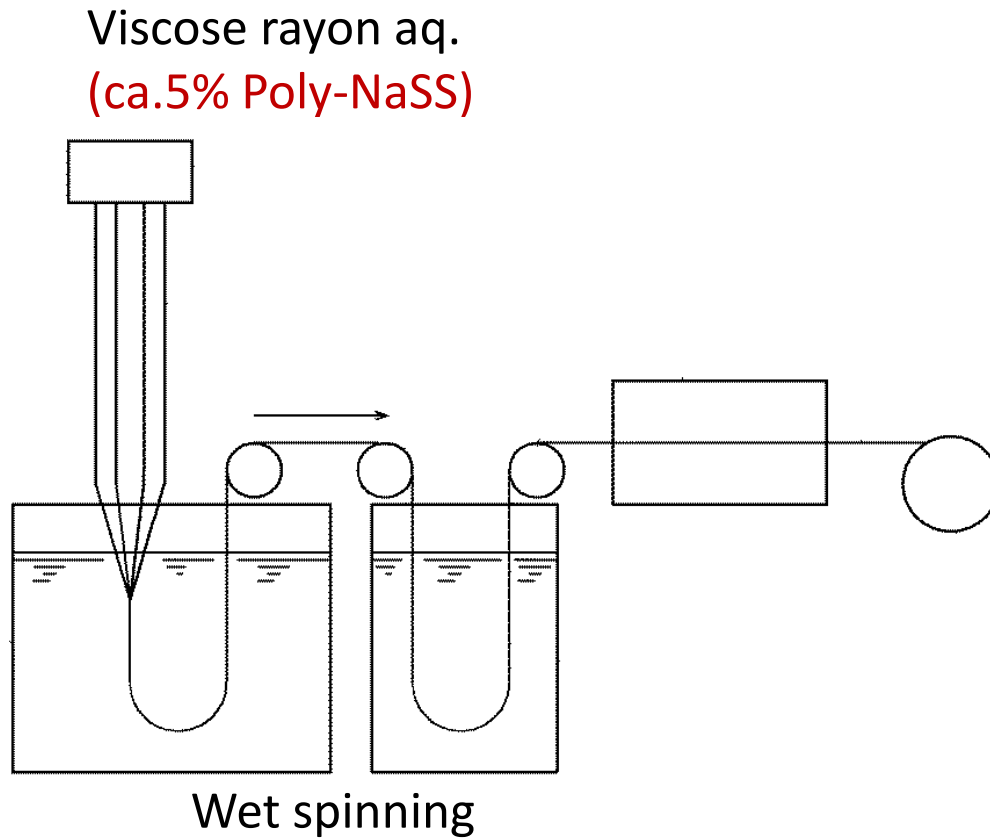
Anti-allergic Fiber

★ Strong electrolyte & Adhesion

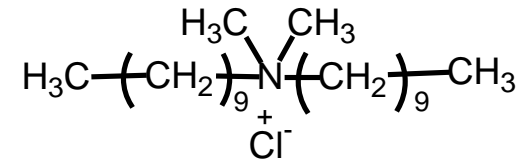


Anti-bacterial Fiber

★ Strong electrolyte & Adhesion



Dip & Dry



Anti-bacterial quaternary ammonium compounds fixed by poly-NaSS

Others



Cation Exchange Membrane



Oil Field Chemicals



Offset Printing



Biomedical Electrode

Oil Field Chemicals for harsh condition

<Application>

- Rheology Modifier
- Scale Inhibitor

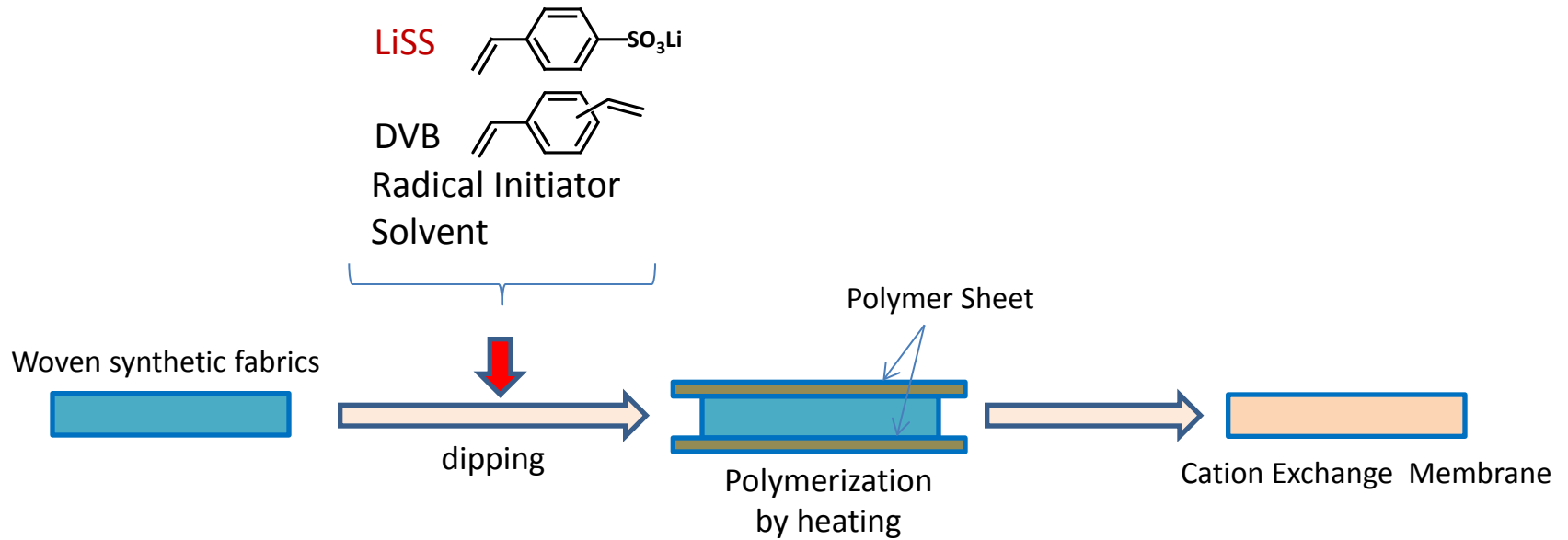
★ Biggest use of AMPS

Table 1 Application of AMPS or ATBS (powder, pellets, aq. solution)

Oil Field Chemicals	EOR(UHMW Poly-AMPS by high purity grade) Fluid loss additive, Scale inhibitor etc
Water Treatment	Scale inhibitor
Textile	Acrylic fiber
Toiletary	Detergent, Skin/Hair care
Coatings	Emulsion paint
Medical	Hydrogel for biomedical electrode
Construction	Concrete formulation, Cement

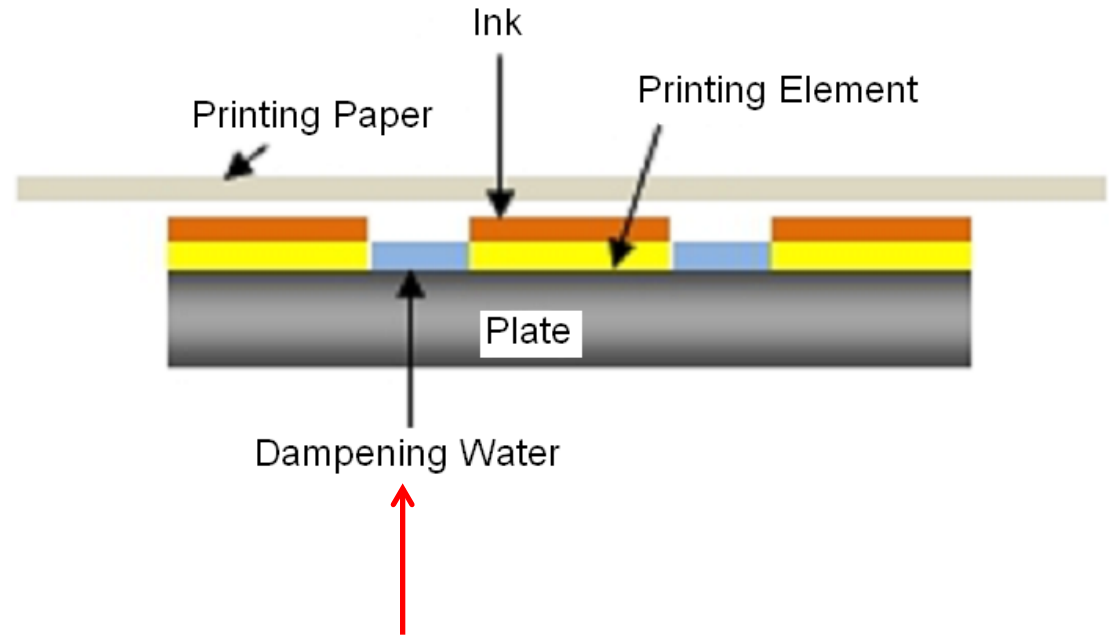
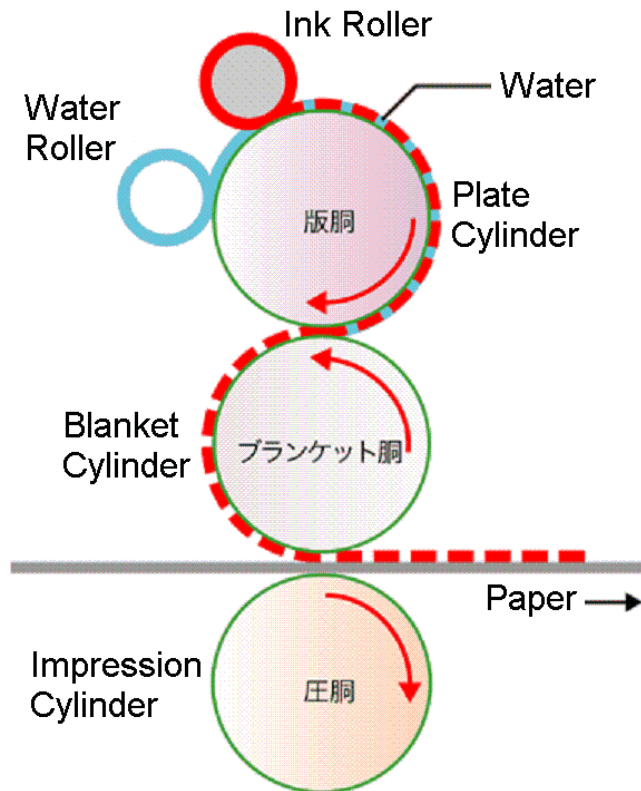
Cation-exchange Membrane

- Simple process without hazardous substances -



Dampening Water of Offset Printing

★ Strong electrolyte & Adhesion

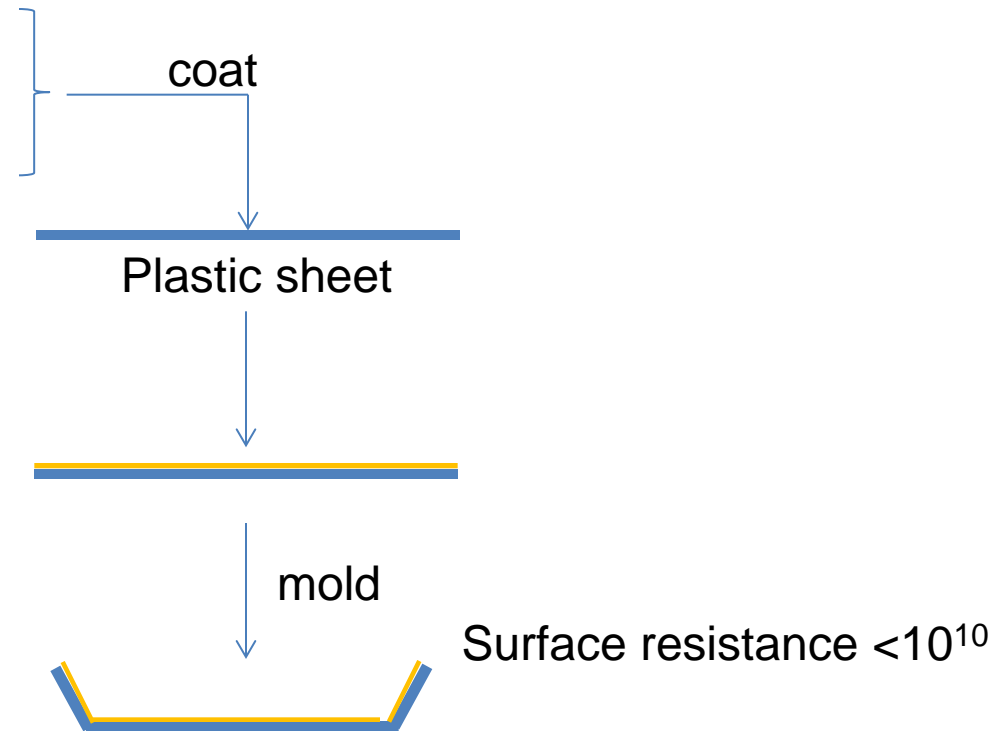


Poly-NaSS is added as an anti-statics

Anti-static plastic tray

★ Strong electrolyte

Water-base binder resin
Poly-NaSS or NaSS monomer aq.
Other additives



Tray for carrying semiconductor

Biomedical Electrode

[JP60-193442,JP3513607]

★ Polymerizability, Low toxicity, Adhesion

NaSS

Water soluble comonomer

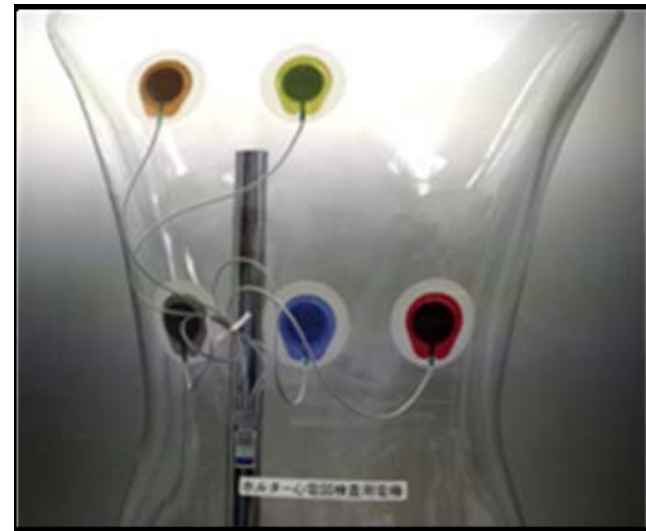
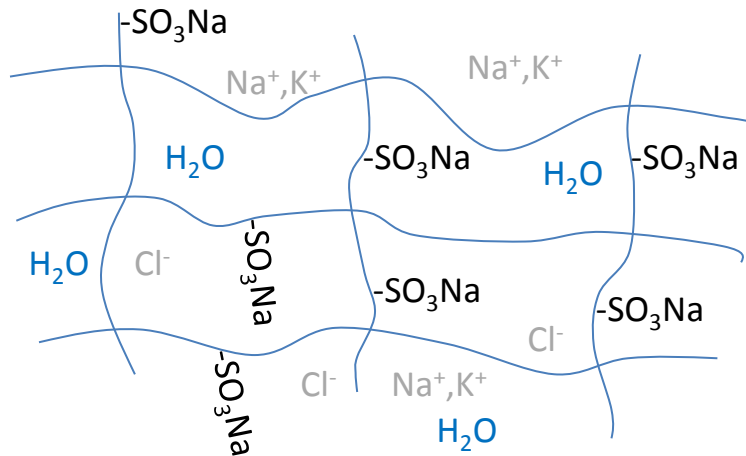
Divinyl monomer

Additives

Water

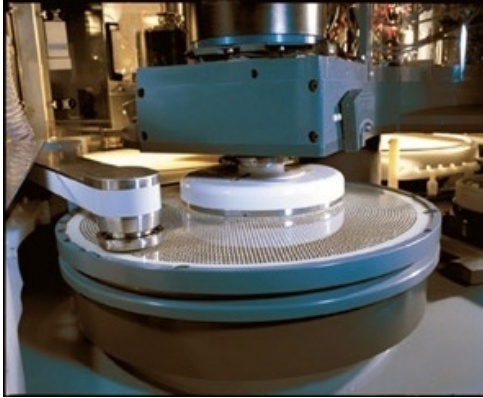


Radical polymerization



Gelatinous electric conductive material

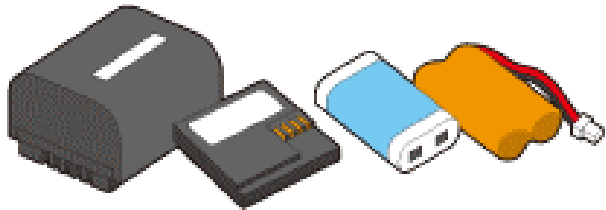
Electronics



CMP



Conductive Polymer

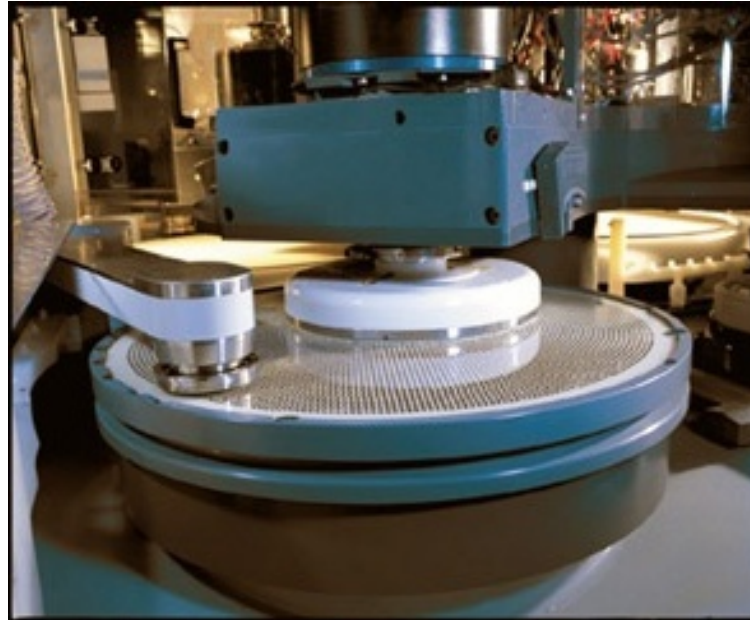


Battery



Plating

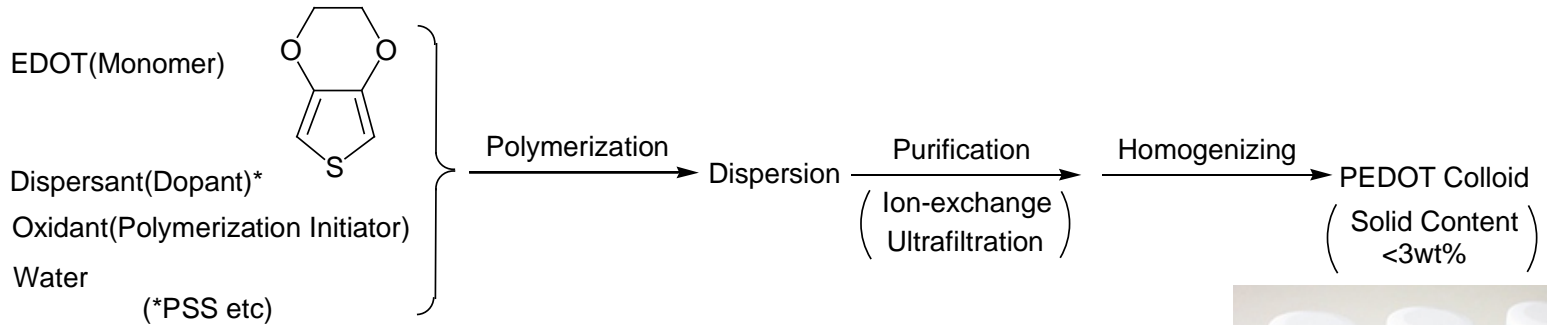
CMP Slurry and Washing Agent



- ★Poly-NaSS acts as a **dispersant** for dust and abrasive grain.
- ★Poly-NaSS prevents scratch and excess abrasion of substrate by **adsorption**.
- ★Poly-NaSS is **easily removed** by washing.
- ★Poly-NaSS is **not foamable**.

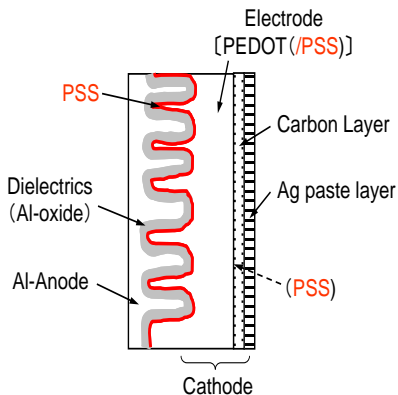
PEDOT/PSS Colloid

★ Strong acid, Rigid chain, Dispersion

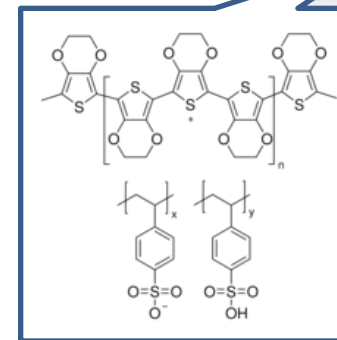


<Example use of Colloid(potential)>

- Transparent Conductive Film
- Aluminum Solid Electrolytic Capacitor

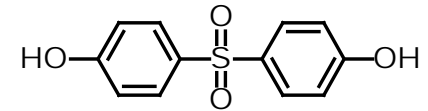
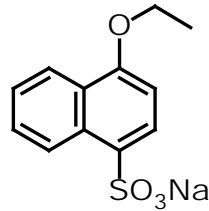
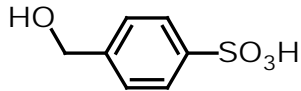
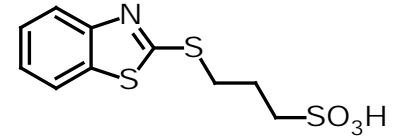
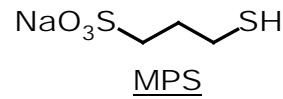
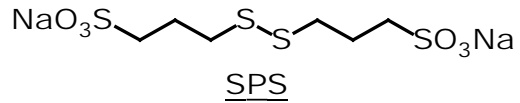


Electric conductivity [S/cm]	Material
100	Electrolyte
10	Electroconductive polymer
1	Charge transfer complex (TCNQ etc)
0.1	Inorganic electrolyte
0.01	MnO ₂
0.005	Electrolytic solution



Plating

Conventional Sulfur Compounds



NaSS and poly-NaSS has been adopted.

